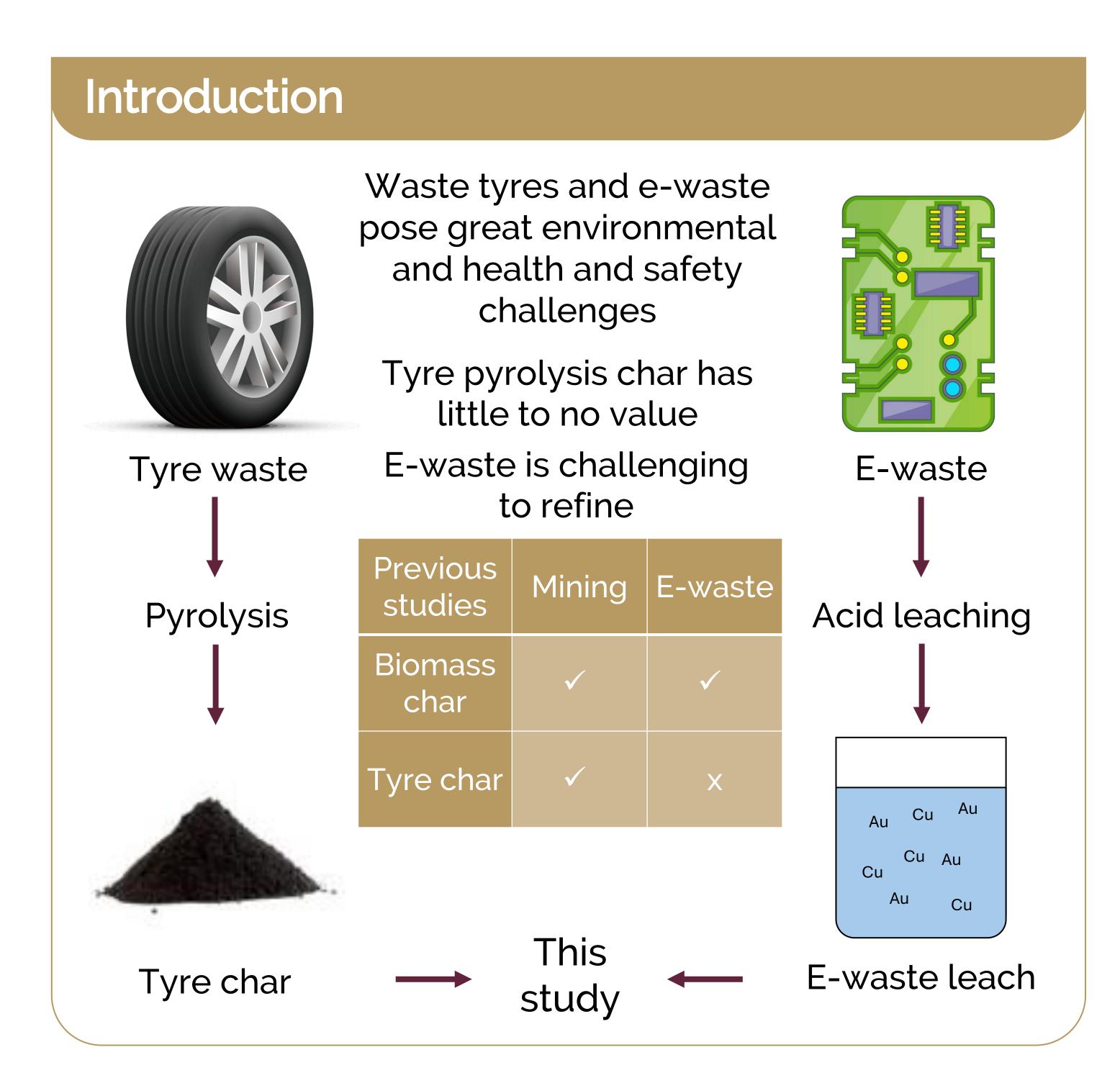


# Valorisation of pyrolysed tyre char as an adsorbent for the recovery of metals from e-waste leach

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# Aim and Objectives

### Aim:

Investigate viability of using tyre char as adsorbent for gold and copper recovery from e-waste leachate.

## Objectives:

- Apply char upgrading methods
- Calculate and compare removal efficiencies of Au and Cu from synthetic e-waste leachate onto tyre char.

# Methodology

### Char pretreatments

- Demineralisation (HCl)
- Chemical (KOH) activation
- Physical (steam) activation

### Synthetic e-waste

- 0.15 g Au(III)/L in 1 M HCl
- 0.15 g Cu(II)/L in 1 M HCl

### Adsorption experiments

- 20 mL e-waste solution
- 20 g char/L
- 40 °C

# **Results and Discussion**

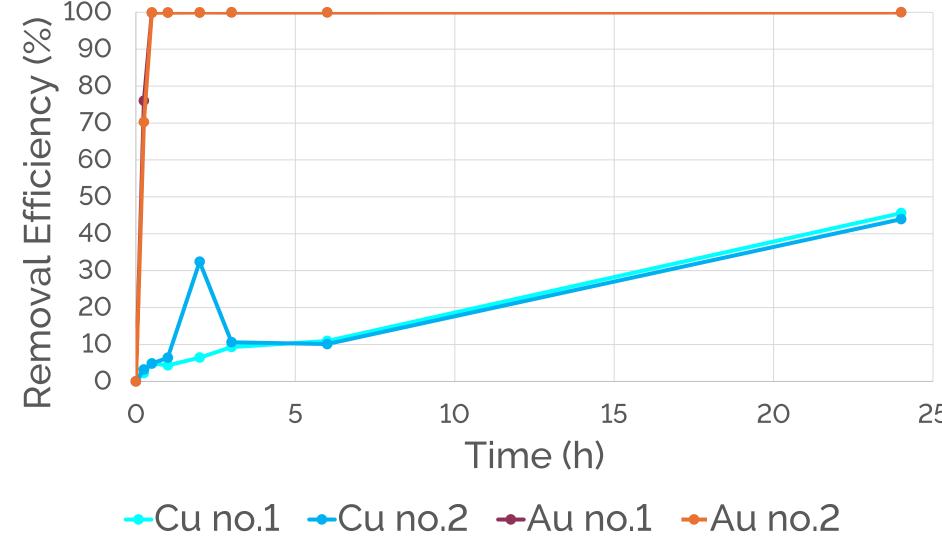


Figure 1: Removal efficiencies for untreated char

Untreated char is suitable for Au recovery, unsuitable for Cu recovery, and unsuitable for separating Au from Cu.

KOH activated char is suitable for Au recovery, unsuitable for Cu recovery, and unsuitable for separating Au from Cu.



Cu no.1 → Cu no.2 → Au no.1 → Au no.2

Time (h)

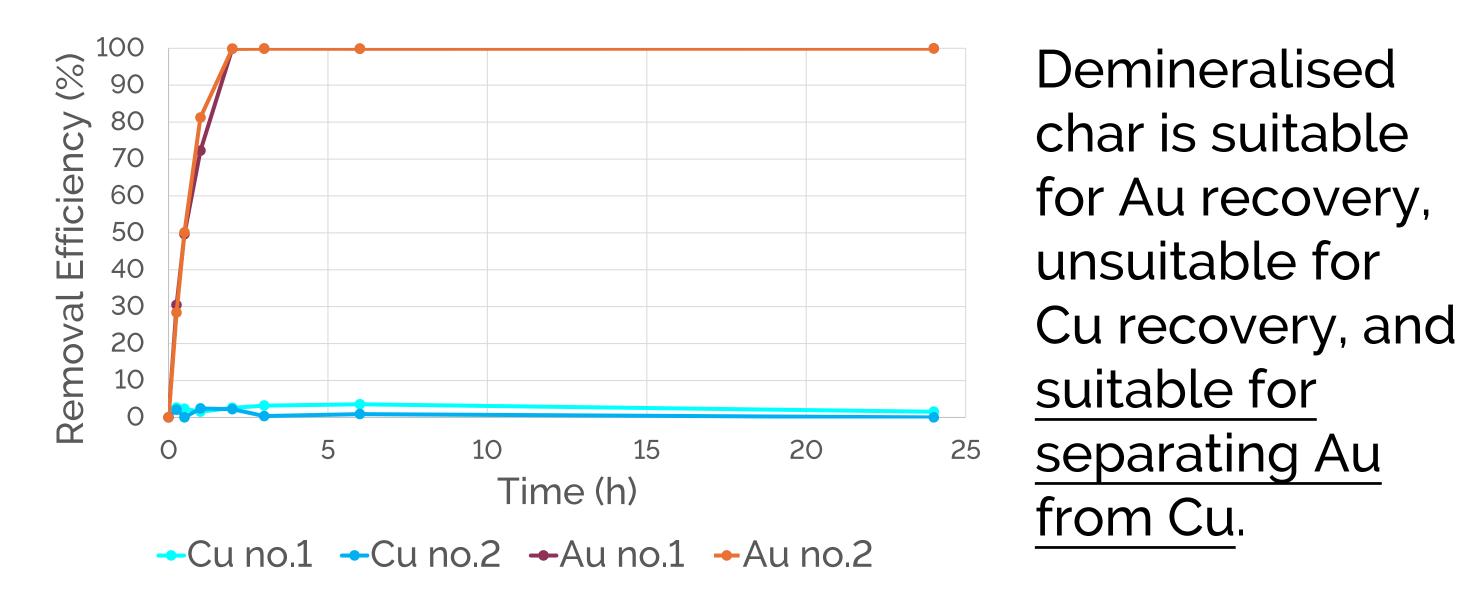
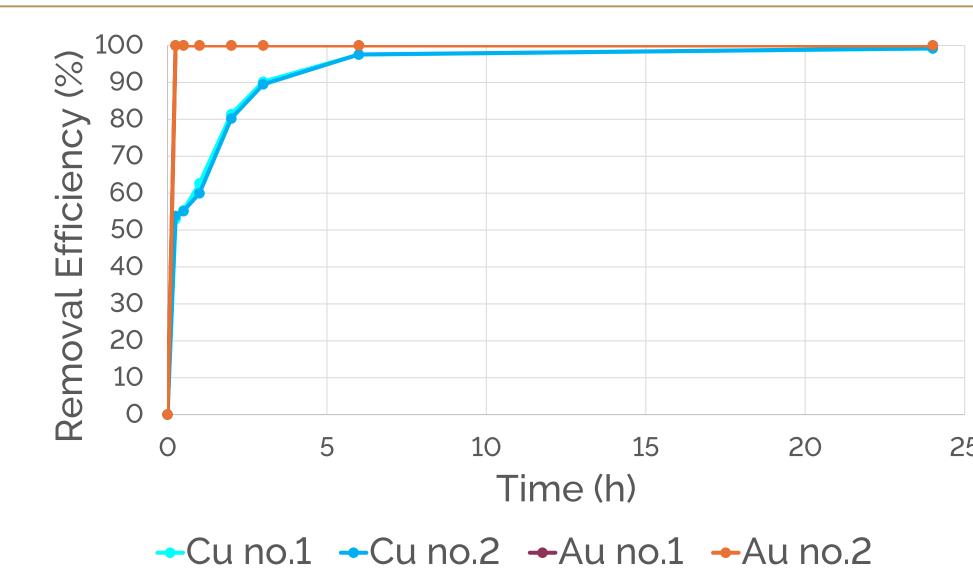


Figure 2: Removal efficiencies for demineralised char



Steam activated char is suitable for Au recovery, suitable for Cu recovery, and unsuitable for separating Au from Cu.

Figure 4: Removal efficiencies for steam activated char

# Conclusions

(%)

Efficiency (

Removal

90

30

- Char upgrading methods of demineralisation, KOH activation and steam activation were applied.
- Demineralisation is the best upgrading method for separating Au from Cu.
- Steam activation is the best upgrading method for recovering both Au and Cu.

# Further work

- Perform temperature trials for kinetic modelling
- Perform char dosage trials for equilibrium modelling
- Test multi-component systems to establish selectivity
- Perform BET surface area analysis to inform results